

1,6,11,18,24,27,33,51,54,60-Decakis(trifluoromethyl)-1,6,11,18,24,27,33,51,54,60-decahydro(C₆₀-I_h)[5,6]fullerene

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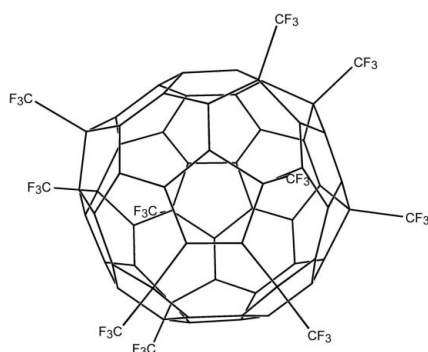
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Key indicators: single-crystal X-ray study; *T* = 296 K; mean $\sigma(\text{C}-\text{C})$ = 0.004 Å; *R* factor = 0.051; *wR* factor = 0.122; data-to-parameter ratio = 11.9.

The title compound, C₇₀F₃₀, is one of six isomers of C₆₀(CF₃)₁₀ that have now been isolated. The fullerene molecule has an idealized I_h C₆₀ core with the ten CF₃ groups arranged in an asymmetric fashion on two ribbons of edge-sharing C₆(CF₃)₂ hexagons, a *para-para-para-meta-para* ribbon and a *para-meta-para* ribbon, giving an overall *p³mp,pmp* structure. There are no cage Csp³–Csp³ bonds. There are intramolecular F···F contacts between pairs of CF₃ groups on the same hexagon that range from 2.568 (3) to 2.682 (3) Å.

Related literature

For related literature, see: Kareev *et al.* (2005); Kareev, Lebedkin, Miller, Anderson, Strauss & Boltalina (2006); Kareev, Lebedkin, Popov, Miller, Anderson, Strauss & Boltalina (2006); Kareev, Shustova, Newell, Miller, Anderson, Strauss & Boltalina (2006); Olmstead *et al.* (2003); Popov *et al.* (2007); Powell *et al.* (2002); Shustova *et al.* (2006); Troyanov *et al.* (2006).



Experimental

Crystal data

C ₇₀ F ₃₀	$\gamma = 65.543 (2)^\circ$
<i>M_r</i> = 1410.70	<i>V</i> = 2281.36 (14) Å ³
Triclinic, <i>P</i> $\bar{1}$	<i>Z</i> = 2
<i>a</i> = 11.0257 (4) Å	Mo <i>K</i> α radiation
<i>b</i> = 11.4172 (4) Å	$\mu = 0.21 \text{ mm}^{-1}$
<i>c</i> = 20.4527 (7) Å	<i>T</i> = 296 (2) K
$\alpha = 82.369 (2)^\circ$	0.20 × 0.15 × 0.07 mm
$\beta = 77.010 (2)^\circ$	

Data collection

Bruker SMART CCD area-detector diffractometer	77789 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2000)	10733 independent reflections
<i>T_{min}</i> = 0.959, <i>T_{max}</i> = 0.986	6553 reflections with <i>I</i> > 2σ(<i>I</i>)
	<i>R_{int}</i> = 0.071

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.051$	901 parameters
$wR(F^2) = 0.122$	$\Delta\rho_{\text{max}} = 0.37 \text{ e \AA}^{-3}$
<i>S</i> = 1.03	$\Delta\rho_{\text{min}} = -0.35 \text{ e \AA}^{-3}$
10733 reflections	

Data collection: *APEX2* (Bruker, 2000); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2158).

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1,6,11,18,24,27,33,51,54,60-Decakis(trifluoromethyl)-1,6,11,18,24,27,33,51,54,60-decahydro(C₆₀-I_h)[5,6]fullerene

N. B. Shustova, D. V. Peryshkov, A. A. Popov, O. V. Boltalina and S. H. Strauss

Comment

Recently reported high-temperature reactions of C₆₀ with CF₃I have yielded five C₆₀(CF₃)₁₀ derivatives, (I)–(V) with thermodynamically stable addition patterns that are asymmetric or dissymmetric as well as unprecedented in fullerene(X)_n chemistry (Kareev *et al.*, 2005; Kareev, Lebedkin, Popov *et al.*, 2006; Kareev, Lebedkin, Miller *et al.*, 2006; Popov *et al.*, 2007). A new member of this set of isomers, the title compound, (VI), has been prepared and we report its crystal structure here.

The structure of (VI), Fig. 1, comprises an idealized I_h C₆₀ core with ten sp³ carbon atoms at positions 1, 6, 11, 18, 24, 27, 33, 51, 54, and 60 (Powell *et al.*, 2002), each of which is attached to a CF₃ group. The core sp³ carbon atoms are not adjacent to one another. The CF₃ groups are arranged on a *para-para-para-meta-para* and *para-meta-para* ribbons of edge-sharing C₆(CF₃)₂ hexagons (*i.e.*, a p³mp,pmp overall addition pattern; see Schlegel diagram in Fig. 1). Note that the shared edges in each ribbon of hexagons are C(sp³)-C(sp²) bonds (*e.g.*, C16—C17, C4—C18, *etc.*), not C(sp²)-C(sp²) bonds. Thus, any pair of adjacent hexagons along the ribbon have a common CF₃ group. As in the recently published structures of three other isomers of C₆₀(CF₃)₁₀ (see below), there are F···F intramolecular contacts between pairs of neighboring CF₃ groups that range from 2.565 (1) to 2.727 (1) Å.

There are now six isomers of C₆₀(CF₃)₁₀ that have been prepared at high temperature, isolated, and characterized. Fluorine-19 NMR spectroscopy has shown that one isomer, (I), has the ten CF₃ groups arranged on a ribbon of seven *meta*- and *para*-C₆(CF₃)₂ edge-sharing hexagons plus an isolated *para*-C₆(CF₃)₂ (Kareev *et al.*, 2005). The other four, C₁-p³mpmpmp-C₆₀(CF₃)₁₀, (II) (Kareev, Lebedkin, Miller *et al.*, 2006), C₁-pmp³mpmp-C₆₀(CF₃)₁₀, (III) (Kareev *et al.*, 2005), C₂-[p³m²(loop)]²-C₆₀(CF₃)₁₀, (IV) (Kareev, Lebedkin, Popov, *et al.*, 2006), and C₁-pmpmpmpmp-C₆₀(CF₃)₁₀, (V) (Popov *et al.*, 2007), have been structurally characterized by single-crystal X-ray diffraction. For comparison, Schlegel diagrams for the six isomers are shown in Fig. 2, arranged according to their DFT relative energies (Popov *et al.*, 2007). The pmp³mpmp ribbon in (III) forms a loop in which two of the *meta*-C₆(CF₃)₂ hexagons have a common C(sp²)-C(sp²) bond (C2—C12). The structure of (IV) is significantly different than the other two isomers in that every CF₃ group has two CF₃ nearest neighbors (*i.e.*, there are no "terminal" CF₃ groups). Instead, it has two symmetry-related p³m² loops of five edge-sharing C₆(CF₃)₂ hexagons that are joined by a C(sp²)-C(sp²) bond that is common to one of the *meta*-C₆(CF₃)₂ hexagons in each loop.

The four shortest cage C—C bonds in (VI) are C4—C5, 1.350 (4) Å, C7—C8, 1.351 (3) Å, C9—C10, 1.359 (3) Å, and C52—C53, 1.348 (4) Å. All four are significantly shorter than the shortest C—C bond in the most precise structure of empty C₆₀ reported to date (C₆₀Pt(octaethylporphyrin)), which is 1.379 (3) Å (Olmstead *et al.*, 2003). More importantly, three

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of these bonds, C4—C5, C9—C10, and C52—C53, are pentagon-hexagon junctions, and the shortest pent-hex junction in $C_{60}\cdot Pt(OEP)$ is 1.440 (3) Å (the longest pent-hex junction in $C_{60}\cdot Pt(OEP)$ is 1.461 (3) Å).

The structure of (VI), predicted to be the most stable isomer of $C_{60}(CF_3)_{10}$, demonstrates a new type of addition pattern for fullerene(CF_3)_n derivatives with $n = 4-12$, two independent ribbons of edge-sharing $C_6(CF_3)_2$ hexagons, to go along with the other six types of addition patterns that have been observed, a single ribbon (*e.g.*, (II)), a ribbon plus an isolated *para*- $C_6(CF_3)_2$ hexagon (Kareev, Shustova, Newell *et al.*, 2006), a single loop of $C_6(CF_3)_2$ hexagons (Trojanov *et al.*, 2006), two loops (*e.g.*, (IV)), a loop plus an isolated hexagon (Shustova *et al.*, 2006), and a loop plus a ribbon (Shustova *et al.*, 2006).

Experimental

The synthesis of (VI) was carried out by heating C_{60} in a stream of CF_3I at 460 °C as previously described (Kareev *et al.*, 2005). Crystals of the HPLC-purified compound were grown by slow evaporation of a saturated benzene solution.

Figures

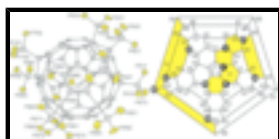


Fig. 1. Left: The molecular structure of (VI) (50% probability ellipsoids; F611 is attached to C61, F621 is attached to C62, *etc.*). Right: Schlegel diagram of (VI), showing the C_{60} core carbon atom numbers (each core carbon atom bearing a CF_3 group is depicted as a black circle) and the p^3mp and pmp ribbons of *meta*- and *para*- $C_6(CF_3)_2$ edge-sharing hexagons (*meta*- $C_6(CF_3)_2$ hexagons are indicated by the letter *m*).

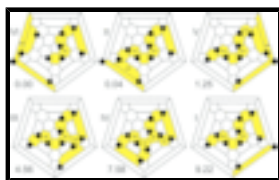


Fig. 2. Schlegel diagrams of (I)–(VI) showing the location of the CF_3 groups as black circles, the IUPAC lowest-locant numbers for the cage carbon atoms to which they are attached, the ribbons or loops of *meta*- and *para*- $C_6(CF_3)_2$ edge-sharing hexagons (*meta*- $C_6(CF_3)_2$ hexagons are indicated by the letter *m*), and the DFT-predicted relative energies.

1,6,11,18,24,27,33,51,54,60-Decakis(trifluoromethyl)-1,6,11,18,24,27,33,51,54,60-decahydro($C_{60}-I_h$)[5,6]fullerene

Crystal data

$C_{70}F_{30}$	$V = 2281.36 (14) \text{ \AA}^3$
$M_r = 1410.70$	$Z = 2$
Triclinic, $P\bar{1}$	$F_{000} = 1380$
Hall symbol: $-P 1$	$D_x = 2.054 \text{ Mg m}^{-3}$
$a = 11.0257 (4) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 11.4172 (4) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$c = 20.4527 (7) \text{ \AA}$	$\mu = 0.21 \text{ mm}^{-1}$
$\alpha = 82.369 (2)^\circ$	$T = 296 (2) \text{ K}$
$\beta = 77.010 (2)^\circ$	Plate, red
$\gamma = 65.543 (2)^\circ$	$0.20 \times 0.15 \times 0.07 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer	10733 independent reflections
Radiation source: fine-focus sealed tube	6553 reflections with $I > 2\sigma(I)$
Monochromator: graphite	$R_{\text{int}} = 0.071$
Detector resolution: 0 pixels mm^{-1}	$\theta_{\text{max}} = 27.9^\circ$
$T = 100(2)$ K	$\theta_{\text{min}} = 2.0^\circ$
φ and ω scans	$h = -14 \rightarrow 14$
Absorption correction: multi-scan (SADABS; Bruker, 2000)	$k = -15 \rightarrow 15$
$T_{\text{min}} = 0.959$, $T_{\text{max}} = 0.986$	$l = -26 \rightarrow 26$
77789 measured reflections	

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.051$	$w = 1/[\sigma^2(F_o^2) + (0.0526P)^2 + 1.012P]$
$wR(F^2) = 0.122$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.03$	$(\Delta/\sigma)_{\text{max}} < 0.001$
10733 reflections	$\Delta\rho_{\text{max}} = 0.37 \text{ e } \text{\AA}^{-3}$
901 parameters	$\Delta\rho_{\text{min}} = -0.35 \text{ e } \text{\AA}^{-3}$
	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.2417 (2)	0.7966 (2)	0.34912 (13)	0.0167 (5)
C2	0.2516 (2)	0.7634 (2)	0.27679 (12)	0.0157 (5)
C3	0.2495 (2)	0.8705 (2)	0.23314 (13)	0.0173 (6)
C4	0.2609 (2)	0.9648 (2)	0.27116 (13)	0.0162 (5)
C5	0.2671 (2)	0.9221 (2)	0.33561 (13)	0.0168 (6)
C6	0.3388 (2)	0.9626 (2)	0.37893 (13)	0.0177 (6)

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C7	0.4466 (2)	0.8404 (2)	0.40600 (12)	0.0169 (6)
C8	0.4474 (2)	0.7213 (2)	0.40838 (12)	0.0162 (5)
C9	0.3574 (2)	0.6916 (2)	0.37692 (12)	0.0162 (5)
C10	0.4272 (3)	0.5707 (2)	0.35399 (12)	0.0162 (5)
C11	0.3963 (3)	0.5230 (2)	0.29636 (13)	0.0176 (6)
C12	0.3211 (2)	0.6414 (2)	0.25287 (13)	0.0162 (5)
C13	0.3861 (3)	0.6252 (2)	0.18461 (13)	0.0175 (6)
C14	0.3740 (2)	0.7309 (2)	0.14025 (12)	0.0166 (6)
C15	0.3069 (2)	0.8550 (2)	0.16543 (13)	0.0167 (6)
C16	0.3800 (2)	0.9347 (2)	0.13014 (13)	0.0166 (6)
C17	0.3899 (2)	1.0259 (2)	0.16422 (13)	0.0168 (6)
C18	0.3072 (2)	1.0677 (2)	0.23469 (13)	0.0178 (6)
C19	0.4165 (2)	1.0723 (2)	0.26848 (13)	0.0169 (6)
C20	0.4315 (3)	1.0228 (2)	0.33214 (13)	0.0189 (6)
C21	0.5635 (3)	0.9656 (2)	0.34854 (13)	0.0173 (6)
C22	0.5765 (3)	0.8555 (2)	0.39425 (12)	0.0180 (6)
C23	0.6973 (3)	0.7512 (2)	0.39117 (12)	0.0177 (6)
C24	0.7061 (3)	0.6125 (2)	0.41016 (13)	0.0184 (6)
C25	0.5740 (2)	0.6062 (2)	0.40545 (12)	0.0169 (6)
C26	0.5631 (3)	0.5161 (2)	0.37195 (12)	0.0168 (5)
C27	0.6828 (3)	0.4085 (2)	0.33531 (13)	0.0185 (6)
C28	0.6564 (3)	0.4076 (2)	0.26429 (13)	0.0183 (6)
C29	0.5305 (3)	0.4583 (2)	0.24692 (13)	0.0175 (6)
C30	0.5150 (3)	0.5120 (2)	0.18045 (13)	0.0172 (6)
C31	0.6259 (3)	0.5114 (2)	0.13233 (12)	0.0173 (6)
C32	0.6174 (3)	0.6223 (2)	0.08498 (12)	0.0174 (6)
C33	0.4824 (2)	0.7288 (2)	0.07731 (12)	0.0165 (5)
C34	0.4910 (3)	0.8593 (2)	0.08202 (12)	0.0168 (6)
C35	0.6130 (3)	0.8730 (2)	0.07202 (12)	0.0160 (5)
C36	0.6250 (3)	0.9616 (2)	0.10971 (13)	0.0166 (5)
C37	0.5142 (3)	1.0392 (2)	0.15430 (13)	0.0173 (6)
C38	0.5328 (3)	1.0666 (2)	0.21882 (13)	0.0173 (6)
C39	0.6600 (3)	1.0195 (2)	0.23429 (13)	0.0176 (6)
C40	0.6768 (3)	0.9647 (2)	0.30016 (13)	0.0189 (6)
C41	0.8032 (3)	0.8540 (2)	0.29596 (13)	0.0188 (6)
C42	0.8110 (3)	0.7479 (2)	0.34053 (13)	0.0186 (6)
C43	0.8836 (2)	0.6200 (2)	0.31604 (13)	0.0182 (6)
C44	0.8164 (3)	0.5385 (2)	0.35195 (13)	0.0192 (6)
C45	0.8045 (2)	0.4494 (2)	0.31806 (13)	0.0191 (6)
C46	0.8588 (2)	0.4371 (2)	0.24768 (13)	0.0178 (6)
C47	0.7705 (3)	0.4108 (2)	0.21537 (13)	0.0183 (6)
C48	0.7565 (2)	0.4596 (2)	0.14998 (13)	0.0168 (6)
C49	0.8269 (3)	0.5382 (2)	0.11544 (13)	0.0180 (6)
C50	0.7365 (3)	0.6373 (2)	0.07528 (12)	0.0165 (6)
C51	0.7523 (2)	0.7629 (2)	0.05412 (13)	0.0179 (6)
C52	0.8333 (2)	0.7879 (2)	0.10011 (13)	0.0174 (6)
C53	0.7617 (2)	0.9049 (2)	0.12601 (12)	0.0169 (6)
C54	0.7919 (3)	0.9537 (2)	0.18355 (13)	0.0189 (6)
C55	0.8700 (3)	0.8365 (2)	0.22661 (13)	0.0185 (6)

C56	0.9363 (2)	0.7156 (2)	0.20313 (13)	0.0183 (6)
C57	0.9411 (2)	0.6057 (2)	0.24838 (13)	0.0187 (6)
C58	0.9285 (2)	0.5115 (2)	0.21303 (13)	0.0188 (6)
C59	0.9144 (2)	0.5601 (2)	0.14573 (13)	0.0190 (6)
C60	0.9458 (2)	0.6816 (2)	0.13085 (13)	0.0192 (6)
C61	0.7442 (3)	0.5715 (3)	0.47912 (14)	0.0239 (6)
C62	0.7206 (3)	0.2761 (3)	0.37207 (14)	0.0234 (6)
C63	0.3201 (3)	0.4339 (3)	0.31551 (13)	0.0226 (6)
C64	0.1010 (3)	0.8192 (3)	0.39247 (13)	0.0204 (6)
C65	0.2467 (3)	1.0509 (3)	0.43616 (14)	0.0230 (6)
C66	0.1864 (3)	1.1957 (2)	0.22957 (13)	0.0203 (6)
C67	0.8628 (3)	1.0456 (3)	0.15876 (14)	0.0247 (6)
C68	1.0902 (3)	0.6488 (3)	0.09045 (13)	0.0222 (6)
C69	0.8119 (3)	0.7730 (3)	-0.02073 (14)	0.0226 (6)
C70	0.4339 (3)	0.7203 (2)	0.01389 (14)	0.0214 (6)
F611	0.76569 (16)	0.44873 (15)	0.49619 (8)	0.0308 (4)
F612	0.85722 (16)	0.58620 (16)	0.48205 (8)	0.0319 (4)
F613	0.64501 (16)	0.64363 (15)	0.52626 (7)	0.0314 (4)
F621	0.74936 (18)	0.27710 (15)	0.43163 (8)	0.0381 (4)
F622	0.62194 (17)	0.23418 (15)	0.38229 (9)	0.0390 (4)
F623	0.82980 (17)	0.18848 (15)	0.33644 (8)	0.0383 (4)
F631	0.39635 (16)	0.31892 (14)	0.33950 (8)	0.0296 (4)
F632	0.20746 (15)	0.48301 (15)	0.36110 (8)	0.0295 (4)
F633	0.28378 (18)	0.41388 (16)	0.26113 (8)	0.0371 (4)
F642	0.06714 (15)	0.72019 (15)	0.39106 (8)	0.0298 (4)
F641	0.09679 (15)	0.83392 (17)	0.45607 (8)	0.0327 (4)
F643	0.00523 (14)	0.92366 (15)	0.37012 (8)	0.0312 (4)
F651	0.19299 (17)	0.98964 (16)	0.48593 (8)	0.0382 (4)
F652	0.31852 (17)	1.09449 (18)	0.46359 (9)	0.0426 (5)
F653	0.14744 (18)	1.15135 (17)	0.41575 (8)	0.0463 (5)
F661	0.12361 (15)	1.24453 (14)	0.28967 (7)	0.0274 (4)
F662	0.22418 (15)	1.28328 (13)	0.19096 (7)	0.0243 (4)
F663	0.09553 (15)	1.17960 (14)	0.20141 (8)	0.0287 (4)
F671	0.97937 (16)	0.99175 (16)	0.11674 (8)	0.0357 (4)
F672	0.78381 (17)	1.15036 (15)	0.12711 (8)	0.0342 (4)
F673	0.88955 (17)	1.08596 (16)	0.21035 (8)	0.0338 (4)
F681	1.11860 (16)	0.75255 (16)	0.07779 (9)	0.0374 (4)
F682	1.18111 (15)	0.56230 (16)	0.12451 (8)	0.0333 (4)
F683	1.10892 (16)	0.59912 (18)	0.03215 (8)	0.0391 (4)
F691	0.92556 (16)	0.67079 (16)	-0.04011 (8)	0.0365 (4)
F692	0.72486 (16)	0.78140 (15)	-0.05850 (7)	0.0290 (4)
F693	0.84067 (17)	0.87724 (16)	-0.03585 (8)	0.0335 (4)
F701	0.52725 (16)	0.70742 (16)	-0.04165 (7)	0.0305 (4)
F702	0.39998 (15)	0.61870 (14)	0.01983 (7)	0.0250 (4)
F703	0.32346 (15)	0.82544 (14)	0.00445 (8)	0.0263 (4)

supplementary materials

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0127 (13)	0.0197 (14)	0.0188 (14)	-0.0072 (11)	-0.0034 (11)	-0.0011 (11)
C2	0.0083 (12)	0.0215 (14)	0.0201 (14)	-0.0083 (11)	-0.0034 (11)	-0.0006 (11)
C3	0.0083 (12)	0.0192 (14)	0.0246 (15)	-0.0046 (10)	-0.0045 (11)	-0.0021 (11)
C4	0.0063 (12)	0.0167 (13)	0.0216 (15)	-0.0015 (10)	0.0008 (10)	-0.0035 (11)
C5	0.0091 (12)	0.0160 (13)	0.0241 (15)	-0.0038 (10)	-0.0021 (11)	-0.0021 (11)
C6	0.0150 (13)	0.0174 (13)	0.0200 (14)	-0.0062 (11)	-0.0008 (11)	-0.0037 (11)
C7	0.0149 (13)	0.0220 (14)	0.0145 (13)	-0.0078 (11)	-0.0013 (11)	-0.0038 (11)
C8	0.0150 (13)	0.0199 (14)	0.0138 (13)	-0.0077 (11)	-0.0013 (10)	-0.0007 (10)
C9	0.0157 (13)	0.0194 (14)	0.0168 (14)	-0.0111 (11)	-0.0025 (11)	0.0017 (11)
C10	0.0186 (14)	0.0161 (13)	0.0158 (13)	-0.0103 (11)	-0.0026 (11)	0.0032 (10)
C11	0.0174 (14)	0.0147 (13)	0.0221 (14)	-0.0079 (11)	-0.0032 (11)	-0.0011 (11)
C12	0.0116 (13)	0.0207 (14)	0.0226 (14)	-0.0118 (11)	-0.0057 (11)	0.0007 (11)
C13	0.0162 (13)	0.0198 (14)	0.0219 (14)	-0.0107 (11)	-0.0070 (11)	-0.0014 (11)
C14	0.0156 (13)	0.0208 (14)	0.0184 (14)	-0.0099 (11)	-0.0077 (11)	-0.0001 (11)
C15	0.0104 (13)	0.0179 (13)	0.0237 (15)	-0.0051 (10)	-0.0075 (11)	-0.0013 (11)
C16	0.0140 (13)	0.0187 (13)	0.0190 (14)	-0.0072 (11)	-0.0080 (11)	0.0042 (11)
C17	0.0127 (13)	0.0143 (13)	0.0206 (14)	-0.0026 (10)	-0.0055 (11)	0.0025 (10)
C18	0.0149 (13)	0.0150 (13)	0.0233 (15)	-0.0052 (11)	-0.0050 (11)	0.0001 (11)
C19	0.0150 (13)	0.0099 (12)	0.0257 (15)	-0.0040 (10)	-0.0041 (11)	-0.0027 (11)
C20	0.0173 (14)	0.0145 (13)	0.0260 (15)	-0.0066 (11)	-0.0013 (11)	-0.0087 (11)
C21	0.0176 (14)	0.0171 (13)	0.0217 (14)	-0.0096 (11)	-0.0046 (11)	-0.0051 (11)
C22	0.0191 (14)	0.0241 (14)	0.0146 (13)	-0.0112 (12)	-0.0027 (11)	-0.0056 (11)
C23	0.0174 (14)	0.0232 (14)	0.0174 (14)	-0.0106 (11)	-0.0081 (11)	-0.0011 (11)
C24	0.0138 (13)	0.0242 (14)	0.0183 (14)	-0.0074 (11)	-0.0053 (11)	-0.0013 (11)
C25	0.0150 (13)	0.0201 (14)	0.0173 (14)	-0.0085 (11)	-0.0058 (11)	0.0030 (11)
C26	0.0175 (14)	0.0177 (13)	0.0153 (13)	-0.0087 (11)	-0.0034 (11)	0.0051 (10)
C27	0.0176 (14)	0.0168 (13)	0.0219 (14)	-0.0075 (11)	-0.0048 (11)	0.0002 (11)
C28	0.0219 (14)	0.0106 (12)	0.0232 (15)	-0.0063 (11)	-0.0071 (12)	0.0003 (10)
C29	0.0216 (14)	0.0117 (13)	0.0226 (15)	-0.0105 (11)	-0.0032 (12)	-0.0008 (11)
C30	0.0186 (14)	0.0140 (13)	0.0225 (14)	-0.0090 (11)	-0.0042 (11)	-0.0033 (11)
C31	0.0208 (14)	0.0151 (13)	0.0175 (14)	-0.0068 (11)	-0.0056 (11)	-0.0041 (10)
C32	0.0184 (14)	0.0172 (13)	0.0179 (14)	-0.0079 (11)	-0.0018 (11)	-0.0053 (11)
C33	0.0151 (13)	0.0181 (13)	0.0161 (13)	-0.0057 (11)	-0.0036 (11)	-0.0013 (10)
C34	0.0179 (14)	0.0174 (13)	0.0159 (14)	-0.0068 (11)	-0.0073 (11)	0.0024 (10)
C35	0.0178 (14)	0.0140 (13)	0.0148 (13)	-0.0062 (10)	-0.0041 (11)	0.0051 (10)
C36	0.0168 (13)	0.0162 (13)	0.0192 (14)	-0.0104 (11)	-0.0033 (11)	0.0038 (10)
C37	0.0180 (14)	0.0124 (13)	0.0216 (14)	-0.0066 (11)	-0.0046 (11)	0.0022 (11)
C38	0.0199 (14)	0.0096 (12)	0.0228 (14)	-0.0068 (11)	-0.0029 (11)	-0.0008 (10)
C39	0.0194 (14)	0.0155 (13)	0.0223 (15)	-0.0112 (11)	-0.0031 (11)	-0.0022 (11)
C40	0.0198 (14)	0.0176 (14)	0.0263 (15)	-0.0122 (11)	-0.0060 (12)	-0.0057 (11)
C41	0.0145 (13)	0.0249 (14)	0.0245 (15)	-0.0129 (11)	-0.0076 (11)	-0.0022 (12)
C42	0.0135 (13)	0.0255 (15)	0.0215 (14)	-0.0101 (11)	-0.0078 (11)	-0.0011 (11)
C43	0.0093 (13)	0.0229 (14)	0.0240 (15)	-0.0061 (11)	-0.0083 (11)	0.0014 (11)
C44	0.0139 (13)	0.0224 (14)	0.0194 (14)	-0.0036 (11)	-0.0082 (11)	0.0016 (11)

C45	0.0119 (13)	0.0175 (13)	0.0243 (15)	-0.0009 (11)	-0.0083 (11)	0.0026 (11)
C46	0.0136 (13)	0.0117 (13)	0.0226 (15)	0.0000 (10)	-0.0035 (11)	0.0006 (11)
C47	0.0178 (14)	0.0091 (12)	0.0248 (15)	-0.0008 (10)	-0.0055 (12)	-0.0026 (11)
C48	0.0163 (13)	0.0129 (13)	0.0197 (14)	-0.0033 (11)	-0.0035 (11)	-0.0037 (10)
C49	0.0150 (13)	0.0174 (13)	0.0187 (14)	-0.0032 (11)	-0.0013 (11)	-0.0053 (11)
C50	0.0178 (14)	0.0162 (13)	0.0142 (13)	-0.0047 (11)	-0.0021 (11)	-0.0048 (10)
C51	0.0142 (13)	0.0201 (14)	0.0201 (14)	-0.0077 (11)	-0.0029 (11)	-0.0012 (11)
C52	0.0137 (13)	0.0224 (14)	0.0178 (14)	-0.0115 (11)	0.0009 (11)	0.0014 (11)
C53	0.0145 (13)	0.0210 (14)	0.0171 (14)	-0.0112 (11)	-0.0004 (11)	0.0021 (11)
C54	0.0159 (13)	0.0193 (14)	0.0251 (15)	-0.0105 (11)	-0.0037 (11)	-0.0016 (11)
C55	0.0136 (13)	0.0255 (15)	0.0233 (15)	-0.0142 (11)	-0.0053 (11)	0.0013 (11)
C56	0.0082 (13)	0.0235 (14)	0.0256 (15)	-0.0079 (11)	-0.0045 (11)	-0.0014 (11)
C57	0.0071 (12)	0.0196 (14)	0.0277 (15)	-0.0027 (10)	-0.0042 (11)	-0.0028 (11)
C58	0.0087 (12)	0.0186 (14)	0.0249 (15)	-0.0004 (11)	-0.0044 (11)	-0.0027 (11)
C59	0.0096 (13)	0.0190 (14)	0.0240 (15)	-0.0025 (11)	0.0004 (11)	-0.0030 (11)
C60	0.0114 (13)	0.0225 (14)	0.0241 (15)	-0.0065 (11)	-0.0036 (11)	-0.0028 (11)
C61	0.0186 (15)	0.0280 (16)	0.0243 (16)	-0.0080 (12)	-0.0062 (12)	0.0013 (12)
C62	0.0253 (15)	0.0199 (14)	0.0247 (16)	-0.0088 (12)	-0.0060 (12)	0.0010 (12)
C63	0.0284 (16)	0.0252 (15)	0.0214 (15)	-0.0177 (13)	-0.0059 (13)	0.0003 (12)
C64	0.0160 (14)	0.0245 (15)	0.0206 (15)	-0.0073 (12)	-0.0049 (11)	-0.0008 (11)
C65	0.0221 (15)	0.0214 (15)	0.0258 (16)	-0.0086 (12)	-0.0038 (12)	-0.0035 (12)
C66	0.0175 (14)	0.0203 (14)	0.0230 (15)	-0.0078 (11)	-0.0038 (12)	-0.0002 (12)
C67	0.0225 (15)	0.0303 (16)	0.0272 (16)	-0.0165 (13)	-0.0034 (13)	-0.0026 (13)
C68	0.0165 (14)	0.0268 (15)	0.0232 (15)	-0.0090 (12)	-0.0035 (12)	0.0008 (12)
C69	0.0187 (14)	0.0215 (15)	0.0257 (16)	-0.0066 (12)	-0.0034 (12)	-0.0008 (12)
C70	0.0201 (14)	0.0199 (14)	0.0260 (16)	-0.0095 (12)	-0.0051 (12)	0.0000 (12)
F611	0.0380 (10)	0.0272 (9)	0.0290 (9)	-0.0125 (8)	-0.0150 (8)	0.0069 (7)
F612	0.0239 (9)	0.0477 (11)	0.0315 (9)	-0.0188 (8)	-0.0144 (7)	0.0040 (8)
F613	0.0297 (9)	0.0385 (10)	0.0221 (9)	-0.0084 (8)	-0.0057 (7)	-0.0049 (7)
F621	0.0600 (12)	0.0261 (9)	0.0295 (10)	-0.0140 (9)	-0.0216 (9)	0.0063 (7)
F622	0.0401 (11)	0.0271 (9)	0.0567 (12)	-0.0209 (8)	-0.0176 (9)	0.0149 (8)
F623	0.0407 (10)	0.0186 (9)	0.0365 (10)	0.0028 (8)	-0.0011 (8)	0.0020 (7)
F631	0.0359 (10)	0.0206 (9)	0.0370 (10)	-0.0171 (8)	-0.0074 (8)	0.0037 (7)
F632	0.0235 (9)	0.0322 (9)	0.0370 (10)	-0.0189 (7)	0.0011 (8)	0.0000 (7)
F633	0.0585 (12)	0.0481 (11)	0.0296 (10)	-0.0434 (10)	-0.0159 (9)	0.0040 (8)
F642	0.0199 (8)	0.0326 (9)	0.0419 (10)	-0.0176 (7)	-0.0008 (7)	-0.0029 (8)
F641	0.0221 (9)	0.0543 (11)	0.0219 (9)	-0.0162 (8)	0.0015 (7)	-0.0086 (8)
F643	0.0131 (8)	0.0317 (9)	0.0407 (10)	-0.0046 (7)	-0.0026 (7)	0.0058 (8)
F651	0.0439 (11)	0.0340 (10)	0.0306 (10)	-0.0176 (8)	0.0139 (8)	-0.0090 (8)
F652	0.0374 (10)	0.0561 (12)	0.0432 (11)	-0.0252 (9)	0.0053 (8)	-0.0320 (9)
F653	0.0426 (11)	0.0359 (11)	0.0331 (10)	0.0149 (8)	-0.0095 (9)	-0.0101 (8)
F661	0.0238 (9)	0.0227 (9)	0.0260 (9)	-0.0008 (7)	-0.0010 (7)	-0.0037 (7)
F662	0.0270 (9)	0.0166 (8)	0.0272 (9)	-0.0066 (7)	-0.0058 (7)	0.0006 (7)
F663	0.0218 (9)	0.0218 (8)	0.0433 (10)	-0.0049 (7)	-0.0153 (8)	-0.0010 (7)
F671	0.0310 (10)	0.0389 (10)	0.0427 (11)	-0.0246 (8)	0.0071 (8)	-0.0078 (8)
F672	0.0416 (10)	0.0272 (9)	0.0425 (10)	-0.0223 (8)	-0.0142 (8)	0.0095 (8)
F673	0.0378 (10)	0.0403 (10)	0.0387 (10)	-0.0293 (9)	-0.0082 (8)	-0.0045 (8)
F681	0.0198 (9)	0.0309 (10)	0.0591 (12)	-0.0140 (8)	0.0033 (8)	0.0004 (8)
F682	0.0128 (8)	0.0401 (10)	0.0383 (10)	-0.0047 (7)	-0.0026 (7)	0.0036 (8)

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F683	0.0230 (9)	0.0627 (12)	0.0328 (10)	-0.0190 (9)	0.0053 (8)	-0.0176 (9)
F691	0.0265 (9)	0.0357 (10)	0.0288 (10)	0.0025 (8)	0.0030 (7)	-0.0042 (8)
F692	0.0293 (9)	0.0423 (10)	0.0204 (9)	-0.0186 (8)	-0.0073 (7)	0.0010 (7)
F693	0.0423 (10)	0.0378 (10)	0.0286 (9)	-0.0283 (9)	-0.0001 (8)	0.0013 (7)
F701	0.0265 (9)	0.0475 (11)	0.0211 (9)	-0.0190 (8)	-0.0014 (7)	-0.0043 (7)
F702	0.0282 (9)	0.0240 (8)	0.0287 (9)	-0.0136 (7)	-0.0088 (7)	-0.0033 (7)
F703	0.0255 (9)	0.0251 (9)	0.0312 (9)	-0.0100 (7)	-0.0131 (7)	0.0024 (7)

Geometric parameters (Å, °)

C1—C9	1.505 (3)	C36—C37	1.388 (3)
C1—C64	1.539 (4)	C36—C53	1.470 (3)
C1—C2	1.545 (3)	C37—C38	1.472 (4)
C1—C5	1.549 (3)	C38—C39	1.373 (4)
C2—C12	1.377 (3)	C39—C40	1.426 (4)
C2—C3	1.410 (4)	C39—C54	1.543 (4)
C3—C15	1.389 (4)	C40—C41	1.437 (4)
C3—C4	1.469 (3)	C41—C42	1.401 (4)
C4—C5	1.350 (4)	C41—C55	1.441 (4)
C4—C18	1.513 (4)	C42—C43	1.438 (4)
C5—C6	1.532 (3)	C43—C57	1.387 (4)
C6—C65	1.533 (4)	C43—C44	1.449 (4)
C6—C7	1.544 (3)	C44—C45	1.368 (4)
C6—C20	1.550 (4)	C45—C46	1.431 (4)
C7—C8	1.351 (3)	C46—C58	1.397 (4)
C7—C22	1.475 (4)	C46—C47	1.435 (4)
C8—C9	1.462 (3)	C47—C48	1.395 (4)
C8—C25	1.464 (3)	C48—C49	1.439 (4)
C9—C10	1.359 (3)	C49—C59	1.373 (4)
C10—C26	1.477 (3)	C49—C50	1.461 (3)
C10—C11	1.519 (3)	C50—C51	1.511 (4)
C11—C63	1.532 (4)	C51—C69	1.534 (4)
C11—C29	1.542 (4)	C51—C52	1.556 (3)
C11—C12	1.546 (3)	C52—C53	1.348 (4)
C12—C13	1.416 (4)	C52—C60	1.519 (4)
C13—C14	1.389 (4)	C53—C54	1.520 (4)
C13—C30	1.466 (3)	C54—C67	1.525 (4)
C14—C15	1.405 (3)	C54—C55	1.543 (4)
C14—C33	1.542 (4)	C55—C56	1.361 (4)
C15—C16	1.473 (4)	C56—C57	1.447 (4)
C16—C17	1.380 (3)	C56—C60	1.548 (4)
C16—C34	1.421 (4)	C57—C58	1.439 (4)
C17—C37	1.408 (4)	C58—C59	1.431 (4)
C17—C18	1.539 (4)	C59—C60	1.541 (4)
C18—C66	1.527 (3)	C60—C68	1.534 (4)
C18—C19	1.539 (3)	C61—F611	1.333 (3)
C19—C20	1.366 (4)	C61—F612	1.339 (3)
C19—C38	1.430 (4)	C61—F613	1.340 (3)
C20—C21	1.425 (4)	C62—F622	1.326 (3)

C21—C40	1.405 (4)	C62—F621	1.328 (3)
C21—C22	1.440 (4)	C62—F623	1.335 (3)
C22—C23	1.366 (4)	C63—F632	1.329 (3)
C23—C42	1.425 (4)	C63—F631	1.331 (3)
C23—C24	1.548 (4)	C63—F633	1.339 (3)
C24—C25	1.512 (3)	C64—F641	1.322 (3)
C24—C61	1.521 (4)	C64—F643	1.333 (3)
C24—C44	1.541 (4)	C64—F642	1.335 (3)
C25—C26	1.367 (3)	C65—F653	1.313 (3)
C26—C27	1.506 (3)	C65—F651	1.326 (3)
C27—C62	1.531 (4)	C65—F652	1.338 (3)
C27—C28	1.546 (4)	C66—F661	1.334 (3)
C27—C45	1.550 (4)	C66—F662	1.335 (3)
C28—C29	1.375 (4)	C66—F663	1.346 (3)
C28—C47	1.431 (4)	C67—F671	1.326 (3)
C29—C30	1.432 (4)	C67—F672	1.339 (3)
C30—C31	1.385 (4)	C67—F673	1.340 (3)
C31—C48	1.423 (3)	C68—F681	1.326 (3)
C31—C32	1.472 (4)	C68—F683	1.326 (3)
C32—C50	1.361 (4)	C68—F682	1.333 (3)
C32—C33	1.509 (3)	C69—F692	1.328 (3)
C33—C70	1.539 (4)	C69—F691	1.333 (3)
C33—C34	1.547 (3)	C69—F693	1.335 (3)
C34—C35	1.386 (4)	C70—F701	1.329 (3)
C35—C36	1.414 (3)	C70—F703	1.339 (3)
C35—C51	1.531 (3)	C70—F702	1.342 (3)
C9—C1—C64	114.0 (2)	C38—C39—C40	118.5 (2)
C9—C1—C2	107.7 (2)	C38—C39—C54	123.6 (2)
C64—C1—C2	111.8 (2)	C40—C39—C54	109.8 (2)
C9—C1—C5	109.8 (2)	C21—C40—C39	120.0 (2)
C64—C1—C5	111.9 (2)	C21—C40—C41	120.1 (2)
C2—C1—C5	100.83 (19)	C39—C40—C41	109.2 (2)
C12—C2—C3	119.2 (2)	C42—C41—C40	118.9 (2)
C12—C2—C1	123.8 (2)	C42—C41—C55	120.9 (2)
C3—C2—C1	108.9 (2)	C40—C41—C55	109.2 (2)
C15—C3—C2	120.8 (2)	C41—C42—C23	120.8 (2)
C15—C3—C4	121.3 (2)	C41—C42—C43	119.3 (2)
C2—C3—C4	108.0 (2)	C23—C42—C43	109.5 (2)
C5—C4—C3	110.7 (2)	C57—C43—C42	118.8 (2)
C5—C4—C18	125.9 (2)	C57—C43—C44	121.1 (2)
C3—C4—C18	120.3 (2)	C42—C43—C44	108.8 (2)
C4—C5—C6	122.8 (2)	C45—C44—C43	119.6 (2)
C4—C5—C1	109.9 (2)	C45—C44—C24	122.8 (2)
C6—C5—C1	123.8 (2)	C43—C44—C24	109.2 (2)
C5—C6—C65	116.1 (2)	C44—C45—C46	119.8 (2)
C5—C6—C7	108.8 (2)	C44—C45—C27	124.1 (2)
C65—C6—C7	110.8 (2)	C46—C45—C27	109.0 (2)
C5—C6—C20	108.5 (2)	C58—C46—C45	121.2 (2)
C65—C6—C20	111.2 (2)	C58—C46—C47	119.2 (2)

supplementary materials

C7—C6—C20	100.25 (19)	C45—C46—C47	109.5 (2)
C8—C7—C22	119.4 (2)	C48—C47—C28	121.0 (2)
C8—C7—C6	123.9 (2)	C48—C47—C46	119.3 (2)
C22—C7—C6	109.5 (2)	C28—C47—C46	109.2 (2)
C7—C8—C9	123.2 (2)	C47—C48—C31	119.5 (2)
C7—C8—C25	121.1 (2)	C47—C48—C49	120.7 (2)
C9—C8—C25	107.2 (2)	C31—C48—C49	107.3 (2)
C10—C9—C8	107.5 (2)	C59—C49—C48	119.7 (2)
C10—C9—C1	125.7 (2)	C59—C49—C50	122.9 (2)
C8—C9—C1	121.4 (2)	C48—C49—C50	107.3 (2)
C9—C10—C26	109.3 (2)	C32—C50—C49	109.2 (2)
C9—C10—C11	123.7 (2)	C32—C50—C51	124.7 (2)
C26—C10—C11	122.6 (2)	C49—C50—C51	121.0 (2)
C10—C11—C63	116.4 (2)	C50—C51—C35	108.0 (2)
C10—C11—C29	108.2 (2)	C50—C51—C69	113.4 (2)
C63—C11—C29	110.7 (2)	C35—C51—C69	110.5 (2)
C10—C11—C12	108.1 (2)	C50—C51—C52	110.5 (2)
C63—C11—C12	111.4 (2)	C35—C51—C52	101.25 (19)
C29—C11—C12	100.8 (2)	C69—C51—C52	112.5 (2)
C2—C12—C13	119.8 (2)	C53—C52—C60	123.2 (2)
C2—C12—C11	123.1 (2)	C53—C52—C51	109.6 (2)
C13—C12—C11	110.0 (2)	C60—C52—C51	123.7 (2)
C14—C13—C12	120.9 (2)	C52—C53—C36	110.6 (2)
C14—C13—C30	120.5 (2)	C52—C53—C54	125.4 (2)
C12—C13—C30	108.7 (2)	C36—C53—C54	120.7 (2)
C13—C14—C15	118.8 (2)	C53—C54—C67	112.0 (2)
C13—C14—C33	123.1 (2)	C53—C54—C39	110.2 (2)
C15—C14—C33	110.6 (2)	C67—C54—C39	110.4 (2)
C3—C15—C14	120.0 (2)	C53—C54—C55	108.4 (2)
C3—C15—C16	120.1 (2)	C67—C54—C55	114.2 (2)
C14—C15—C16	108.4 (2)	C39—C54—C55	101.0 (2)
C17—C16—C34	121.0 (2)	C56—C55—C41	119.8 (2)
C17—C16—C15	119.9 (2)	C56—C55—C54	122.6 (2)
C34—C16—C15	108.8 (2)	C41—C55—C54	109.1 (2)
C16—C17—C37	119.6 (2)	C55—C56—C57	119.6 (2)
C16—C17—C18	123.3 (2)	C55—C56—C60	124.3 (2)
C37—C17—C18	110.4 (2)	C57—C56—C60	109.1 (2)
C4—C18—C66	111.0 (2)	C43—C57—C58	119.1 (2)
C4—C18—C17	110.8 (2)	C43—C57—C56	121.5 (2)
C66—C18—C17	110.1 (2)	C58—C57—C56	108.8 (2)
C4—C18—C19	108.5 (2)	C46—C58—C59	120.9 (2)
C66—C18—C19	115.2 (2)	C46—C58—C57	119.2 (2)
C17—C18—C19	100.76 (19)	C59—C58—C57	109.0 (2)
C20—C19—C38	120.0 (2)	C49—C59—C58	119.9 (2)
C20—C19—C18	122.4 (2)	C49—C59—C60	123.5 (2)
C38—C19—C18	109.9 (2)	C58—C59—C60	109.9 (2)
C19—C20—C21	119.6 (2)	C52—C60—C68	115.8 (2)
C19—C20—C6	124.6 (2)	C52—C60—C59	109.4 (2)
C21—C20—C6	109.4 (2)	C68—C60—C59	110.6 (2)

C40—C21—C20	120.1 (2)	C52—C60—C56	109.2 (2)
C40—C21—C22	119.3 (2)	C68—C60—C56	110.4 (2)
C20—C21—C22	110.7 (2)	C59—C60—C56	100.3 (2)
C23—C22—C21	120.7 (2)	F611—C61—F612	107.2 (2)
C23—C22—C7	121.5 (2)	F611—C61—F613	107.5 (2)
C21—C22—C7	106.9 (2)	F612—C61—F613	107.4 (2)
C22—C23—C42	120.1 (2)	F611—C61—C24	112.7 (2)
C22—C23—C24	122.5 (2)	F612—C61—C24	111.5 (2)
C42—C23—C24	109.6 (2)	F613—C61—C24	110.3 (2)
C25—C24—C61	113.7 (2)	F622—C62—F621	107.3 (2)
C25—C24—C44	108.4 (2)	F622—C62—F623	107.0 (2)
C61—C24—C44	114.4 (2)	F621—C62—F623	107.3 (2)
C25—C24—C23	109.3 (2)	F622—C62—C27	112.3 (2)
C61—C24—C23	109.4 (2)	F621—C62—C27	111.9 (2)
C44—C24—C23	100.8 (2)	F623—C62—C27	110.7 (2)
C26—C25—C8	107.7 (2)	F632—C63—F631	107.8 (2)
C26—C25—C24	125.0 (2)	F632—C63—F633	107.3 (2)
C8—C25—C24	122.1 (2)	F631—C63—F633	107.0 (2)
C25—C26—C10	108.2 (2)	F632—C63—C11	112.6 (2)
C25—C26—C27	123.9 (2)	F631—C63—C11	112.2 (2)
C10—C26—C27	123.4 (2)	F633—C63—C11	109.8 (2)
C26—C27—C62	115.3 (2)	F641—C64—F643	108.2 (2)
C26—C27—C28	107.9 (2)	F641—C64—F642	107.1 (2)
C62—C27—C28	112.6 (2)	F643—C64—F642	107.0 (2)
C26—C27—C45	107.9 (2)	F641—C64—C1	111.9 (2)
C62—C27—C45	111.2 (2)	F643—C64—C1	111.0 (2)
C28—C27—C45	100.8 (2)	F642—C64—C1	111.4 (2)
C29—C28—C47	119.1 (2)	F653—C65—F651	108.2 (2)
C29—C28—C27	124.9 (2)	F653—C65—F652	107.7 (2)
C47—C28—C27	109.2 (2)	F651—C65—F652	105.5 (2)
C28—C29—C30	120.1 (2)	F653—C65—C6	112.5 (2)
C28—C29—C11	124.1 (2)	F651—C65—C6	112.3 (2)
C30—C29—C11	109.4 (2)	F652—C65—C6	110.2 (2)
C31—C30—C29	121.0 (2)	F661—C66—F662	108.0 (2)
C31—C30—C13	119.6 (2)	F661—C66—F663	108.0 (2)
C29—C30—C13	108.7 (2)	F662—C66—F663	107.0 (2)
C30—C31—C48	119.3 (2)	F661—C66—C18	111.9 (2)
C30—C31—C32	121.4 (2)	F662—C66—C18	111.6 (2)
C48—C31—C32	107.8 (2)	F663—C66—C18	110.1 (2)
C50—C32—C31	108.4 (2)	F671—C67—F672	107.8 (2)
C50—C32—C33	124.9 (2)	F671—C67—F673	107.4 (2)
C31—C32—C33	121.2 (2)	F672—C67—F673	107.2 (2)
C32—C33—C70	113.3 (2)	F671—C67—C54	112.7 (2)
C32—C33—C14	110.3 (2)	F672—C67—C54	110.8 (2)
C70—C33—C14	110.1 (2)	F673—C67—C54	110.8 (2)
C32—C33—C34	108.4 (2)	F681—C68—F683	107.7 (2)
C70—C33—C34	113.5 (2)	F681—C68—F682	107.7 (2)
C14—C33—C34	100.45 (19)	F683—C68—F682	107.0 (2)
C35—C34—C16	119.2 (2)	F681—C68—C60	111.1 (2)

supplementary materials

C35—C34—C33	122.8 (2)	F683—C68—C60	112.5 (2)
C16—C34—C33	109.7 (2)	F682—C68—C60	110.6 (2)
C34—C35—C36	119.6 (2)	F692—C69—F691	106.8 (2)
C34—C35—C51	124.1 (2)	F692—C69—F693	107.2 (2)
C36—C35—C51	108.8 (2)	F691—C69—F693	107.5 (2)
C37—C36—C35	120.7 (2)	F692—C69—C51	111.0 (2)
C37—C36—C53	121.0 (2)	F691—C69—C51	112.4 (2)
C35—C36—C53	108.1 (2)	F693—C69—C51	111.7 (2)
C36—C37—C17	119.7 (2)	F701—C70—F703	107.5 (2)
C36—C37—C38	120.0 (2)	F701—C70—F702	107.2 (2)
C17—C37—C38	108.8 (2)	F703—C70—F702	107.1 (2)
C39—C38—C19	121.6 (2)	F701—C70—C33	113.2 (2)
C39—C38—C37	120.4 (2)	F703—C70—C33	111.1 (2)
C19—C38—C37	108.2 (2)	F702—C70—C33	110.4 (2)

Fig. 1

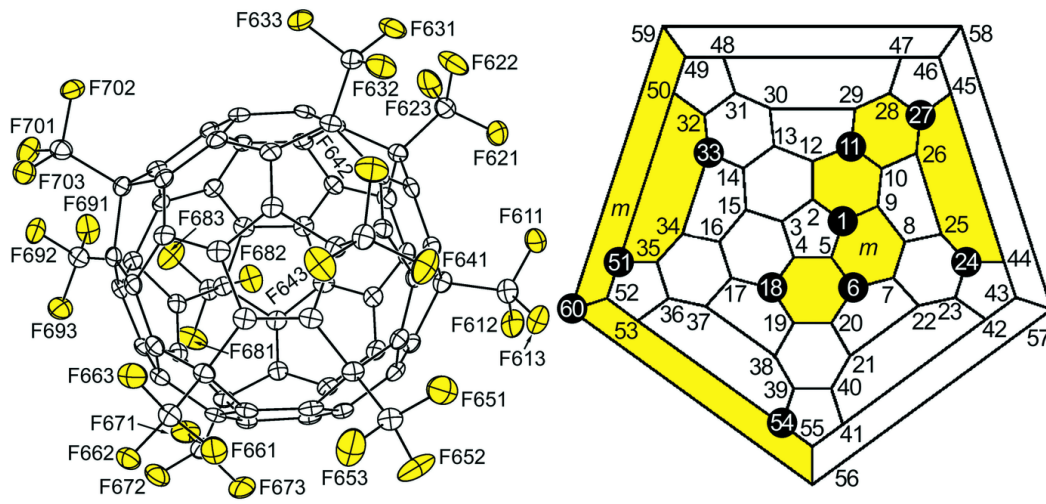


Fig. 2

